(Autonomous Institution - UGC, Govt. of India) UG Model question paper Engineering Chemistry

	Engineering Chemistry	
I	Time: 3 hours Note: This question paper contains of 5 sections. Answer five questions, choosing one question from each section and each question carries 14 marks.	
	Section-I	
]	 a) What are Reference electrodes? Explain the construction and working of Saturated Colomel electrode with the help of a neat sketch. b) Derive Nernst equation and its applications c) Define conductance, equivalent and molar conductance. Give their units. 	[7M] [4M] [3M]
	OR	
2	a) Sort out the differences between primary and secondary batteries b)Explain the construction and working of Lead-Acid storage cell with neat sketch. c) Explain the construction and working of Alkaline fuel cell with neat sketch.	[2M] [6M] [6M]
	Section-II	
3	a)Explain the various factors affecting corrosion	[7M]
	b) What is electrochemical corrosion. Explain the mechanism of electrochemical corrosi by evolution of hydrogen and absorption of oxygen?	
	OR	
4	 a) Define Cathodic protection. Explain sacrificial anodic protection and impressed curre cathodic protection method with applications b)Define Hot Dipping. Explain the process of galvanization and Tinning with the help of a neat sketch 	[7M]
	Section-III	[, , -]
Ţ	 a) Define natural rubber ,classification .Explain the process of vulcanization of rubber b) Explain the preparation, properties and applications of Bakelite, Teflon and polyethy 	[7M] lene? [7M]
	OR	
(6 a)What are conducting polymers. Explain the classification and mechanism of conducting polymers	ng [7M]
	b) Write a note on(i) flash and firepoint (ii)cloud and pour point	[7M]
	Section-IV	
7	7 a) Write a note on (i) Scales and sludges (ii) Caustic embrittlement (iii) priming and foaming	[7M]
	b)Explain zeolite process with the help of a neat sketch	[7M]
	OR	
8	8 a) What is desalination of brackish water? Explain Reverse Osmosis and its significance.	[4M]
•	b) Explain about break point chlorination	[4M]
	a) Francis I on Freshange and good with the help of a next distab	

[6M]

c)Explain Ion Exchange process with the help of a neat sketch.

Section-V

9 a) Explain the proximate analysis of coal with its significance. [7M] b).Define petroleum.Explain refining of petroleum by fractional distillation [7M]

OR

- 10 a) Define cracking. Explain the process of fixed bed catalytic cracking with a neat sketch. [7M]
 - b) Define calorific value ,HCV,LCV. Explain the calorific value of gaseous fuel by Junkers gas calorimeter [7M]

(Autonomous Institution – UGC, Govt. of India) UG Model question paper Engineering Chemistry

Time: 3 hours Max Marks: 70

Note: This question paper contains of 5 sections. Answer five questions, choosing one question from each section and each question carries 14 marks.

CECTION I		
SECTION-I 1 a) Explain the construction and working of calomel electrode with neat sketch.	[6M]	
b) Explain the construction and working of Galvanic cell.	[4M]	
c) Define specific, molar and equivalent conductance. Give their units.	[4M]	
OR		
2 a) Explain the construction and working of Lead-Acid storage cell with neat sketch	ı [9M]	
b) Explain the construction and working of H_2 - O_2 fuel cell with neat sketch. SECTION-II	[5M]	
3 a) What is electrochemical corrosion. Explain the mechanism of electrochemical		
corrosion by evolution of hydrogen and absorption of oxygen?	[10M]	
b) Define Galvanising, Tinning and Metal Cladding with examples.	[4M]	
OR	bosse	
4 a) What is cathodic protection. Explain both sacrificial anodic protection and improdurent cathodic protection method?	[14M]	
SECTION-III	[14M]	
5 a) Explain the Zeolite process for softening of hard water?	[9M]	
b) Write a short note on Scales and Sludges.	[5M]	
OR		
6 a) Explain the Reverse Osmosis and its significance.	[4M]	
b) Explain break point chlorination with a neat sketch.	[5M]	
c) Give a brief note on phosphate, calgon and colloidal conditioning.	[5M]	
SECTION-IV 7 a) Explain the proximate analysis of coal with its significance.	[MO]	
b) Explain the ultimate analysis of coal with respect to nitrogen.	[9M] [5M]	
OR	[JM]	
8 a) Define cracking. Explain the process of fixed bed catalytic cracking with a neat s	ketch.	
	[8M]	
b) Define Calorific value, HCV and LCV of a fuel.	[3M]	
c) Give a brief note on octane and cetane number.	[3M]	
SECTION-V		
9 a) Difference between Addition polymerization and Condesation polymerization?	[5M]	
b) Explain the preparation, properties and applications of Bakelite, Teflon and		
polyethylene?	[9M]	
OR		
10 a)What are conducting polymers. Explain the mechanism of conduction in	[O] 43	
polyacetylene?	[9M]	
b) write a short note on Biodegradable polymers.		

(Autonomous Institution - UGC, Govt. of India) UG Model question paper Engineering Chemistry

Time: 3 hours Max Marks: 70

Note: This question paper contains of 5 sections. Answer five questions, choosing one question from each section and each question carries 14 marks.

Section-I	LOMI	
1 a) Explain the construction and working of Lead-Acid storage cell with neat sketch b) Explain the construction and working of H ₂ -O ₂ fuel cell with neat sketch. OR	[5M]	
2 a) Explain the construction and working of calomel electrode with neat sketch.	[6M]	
b) Explain the construction and working of Galvanic cell.	[4M]	
c) Define specific, molar and equivalent conductance. Give their units.	[4M]	
Section-II		
3 a) what is cathodic protection. Explain both sacrificial anodic protection and impressed		
current cathodic protection method?	[14M]	
OR		
4 a) what is electrochemical corrosion. Explain the mechanism of electrochemical		
corrosion by evolution of hydrogen and absorption of oxygen?	[10M]	
b) Define Galvanising, Tinning and Metal Cladding with examples.	[4M]	
Section-III		
5 a) Explain the Reverse Osmosis and its significance.	[4M]	
b) Explain break point chlorination with a neat sketch.	[5M]	
c) Give a brief note on phosphate, calgon and colloidal conditioning. OR	[5M]	
6 a) Explain the Zeolite process for softening of hard water?	[9M]	
b) Write a short note on Scales and Sludges.	[5M]	
Section-IV		
7 a) Explain the proximate analysis of coal with its significance.	[9M]	
b) Explain the ultimate analysis of coal with respect to nitrogen.	[5M]	
OR		
8 a) Define cracking. Explain the process of fixed bed catalytic cracking with a neat		
	[8M]	
b) Define Calorific value, HCV and LCV of a fuel.	[3M]	
c) Give a brief note on octane and cetane number.	[3M]	
Section-V		
9 a)What are conducting polymers. Explain the mechanism of conduction in	[0] [1]	
polyacetylene?	[9M]	
b) write a short note on Biodegradable polymers. OR	[5M]	
10 a) Difference between Addition polymerization and Condesation polymerization	2[5M]	
b) Explain the preparation, properties and applications of Bakelite, Teflon and	[מוען]	
polyethylene?	[9M]	
polyethylene.	[71,1]	

(Autonomous Institution – UGC, Govt. of India) UG Model question paper Engineering Chemistry

Time: 3 hours Max Marks: 70

Note: This question paper contains of 5 sections. Answer five questions, choosing one question from each section and each question carries 14 marks.

Answer all the questions

SECTION-I

1.	a) Explain the construction of calomel electrode.b) Define equivalent conductance and its unit.c) Define galvanic series.d) Explain galvanic cell.	[7M] [2M] [2M] [3M]
2.	a) Explain the construction of hydrogen oxygen fuel cell.b) Define secondary batteries with Lead acid cells.c) Derive Nernst equation.	[7M] [4M] [3M]
	SECTION-II	
3.	 a). Explain electrochemical corrosion. b) Write about factors affecting rate of corrosion: (i) Temperature (ii) Passivity (iii) Humidity c) Write short notes on: (i) Stable layer (ii) Unstable layer(iii) Volatile layer (iv) Porous layer OR 	[7M] [3M] [4M]
4.	a) Explain the process of galvanizing and tinning.b) Describe the method of cathodic protection .	[7M] [7M]
SECTION-III		
5.	 a) Write a note on fabrication of plastics. b) Define elastomers. Give the preparation and application of (i) Buty rubber (ii) Buna-S c) Define conducting polymer d) Explain addition polymerization OR 	[4M] [4M] [2M] [3M]
6.	a) Explain the difference between thermoplastics & thermosetting plastics.b) Outline the preparation properties and uses of	[4M] [4M]

	(i) Bakelite (ii) Nylon-6,6 (iii) PVC c) Define condensation polymerization d) Explain compounding of plastics SECTION-IV	[2M] [3M]
7.	 a) Explain break point chlorination. b) Describe the lime soda process for softening of hard water. c) Explain reverse osmosis d) Write short note on: (i) Scales (ii) Colloidal Conditioning (iii) Calgon Conditioning OR 	[4M] [4M] [2M] [3M]
8.	 a) Explain softening of water by zeolite process. b) Define caustic embrittlement with reaction. How can this be prevented? c) Write short note on: (i) Priming (ii) Phosphate Conditioning (iii) Sludge d) Explain sterilization of water by chlorination. 	[4M] [4M] [3M]
SECTION-V		
9.	a) Describe the Fischer Tropsch's process of synthetic petrol.b) Define cracking. Discuss any one method of catalytic cracking.c) Define octane number	[5M] [5M] [4M]
10	 a) Explain knocking in internal combustion engine & prevention of knocking b) Define petroleum. How is it refined by fractional distillation? Write various fractions with boiling range. c) Define cetane number 	

(Autonomous Institution – UGC, Govt. of India) UG Model question paper Engineering Chemistry

Time: 3 hours Max Marks: 70

Note: This question paper contains of 5 sections. Answer five questions, choosing one question from each section and each question carries 14 marks.

SECTION-I	[0][
1 a) Write Nernst equation and give its applications.	[3M]
b) Give a detailed account on the Lead – Acid cell with appropriate chemical reaction of Explain the construction and working of a Calvania cell	
c)Explain the construction and working of a Galvanic cell. OR	[5M]
2 a) Explain the construction and functioning of H_2 - O_2 fuel cell. Give the advantages	and
applications of fuel cells.	[6M]
b)What is reference electrode. Explain the construction and working of Glass elect	
,	[6M]
c) What is an electrochemical series. Write two important applications.	[2M]
SECTION-II	
3 a)Explain rusting of iron with the help of electrochemical theory of corrosion.	[7M]
b)Explain the processes of Galvanising and Tinning.	[7M]
OR	
4 a) What is cathodic protection. Explain methods of cathodic protection.	[6M]
b) Discuss how nature of metal and nature of environment affect the rate of corros	
	[4M]
c) Explain Electroless plating.	[4M]
SECTION-III	
5 a) Differentiate thermoplatic resins and thermoset resins.	[2M]
b)Write the preparation properties and applications of PVC,Teflon and Bakelite.	[7M]
c)Give the characteristics of a good lubricant and explain flash and fire points. OR	[5M]
6 a)What are conducting polymers. Explain the mechanism of conduction in	
polyacetylene.	[6M]
b)Explain vulcanization of rubber. Write the preparation, properties and applicati	
buns-S butyl rubber.	[5M]
c) Distinguish between addition and condensation polymerization.	[3M]
SECTION-IV	
7 a)Write a note on caustic embrittlement, priming and foaming.	[7M]
b)How municipal water is disinfected by chlorination and ozonisation. Explain bre	-
chlorination.	[7M]
OR	
8 a) Explain Zeolite process for softening of water. How exhausted zeolites are regenerated.	[7M]
b) Explain how scales and sludges are formed in boilers. Write about Phosphate an	

[7M]

conditioning for prevention of scales.

SECTION-V

9 a) Define HCV and LCV.	[3M]
b) Explain ultimate analysis and give its significance.	[6M]
c) Describe with a neat sketch the manufacturing of Fisher –Tropsch's process.	[5M]
OR	
10 a) What is cracking. Explain in detail about fixed bed catalytic cracking with a neat	
sketch.	[6M]
b) Give brief note on octane and cetane rating.	[6M]
c)What are the characteristics of a good fuel.	[2M]